

Emerging Nanotechnology-based Corrosion Control Coatings

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Emerging Nanotechnology-based Corrosion Control Coatings

Outline:

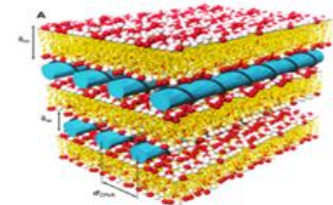
- ◆ The Impact of nanotechnology
- ◆ Application in Corrosion Control Coatings

Nanotechnology

Nanotechnology – the use of nano-sized materials to produce macro-sized products

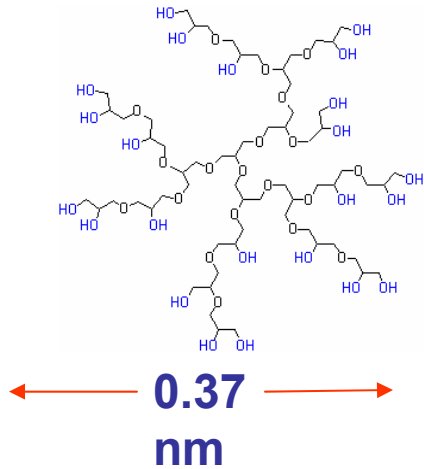
- The problem with this definition is that most of chemistry, materials physics and a sizeable fraction of materials engineering and biochemistry would fall within this definition

Nanoscience is being touted as the engine that will drive the next industrial revolution

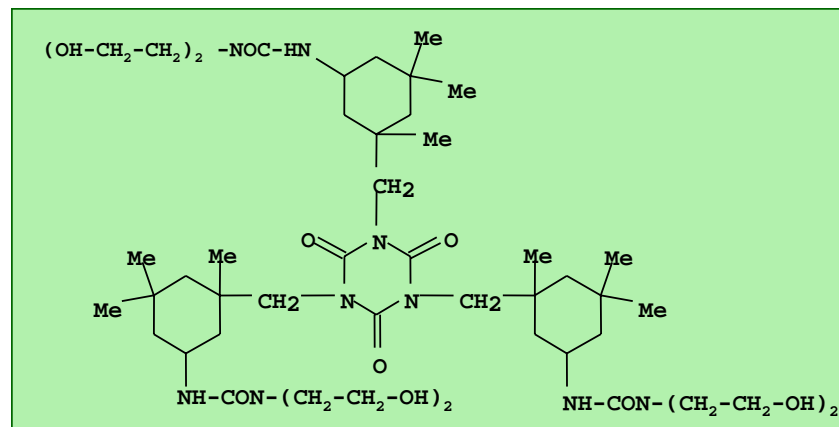


Nano-scale and Conventional Materials

Latex particle size	10 – 1000 nm
TiO ₂ pigment particle	200 – 500 nm
Polyurethane dispersion	50 – 100 nm
Dissolved polymers	2 – 100 nm
Organic molecules	0.2 – 5 nm

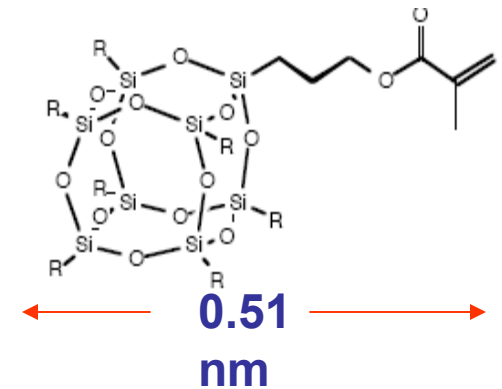


Hydroxyl functional
Polyether dendrimer



Hydroxyl functional IPDI-based dendrimer

J. Baghdachi, *et al*



Methacrylate functional
silsesquioxane

Nano-materials

Aluminum oxide

Barium oxide

Carbon black

Calcium carbonate

Carbon nanotubes

Cerium oxide

Dendrimers, hyperbranched and supramolecules

Indium tin oxide

Nano-clays

Organic polymers

Silicone dioxide

Titanium dioxide

Zinc oxide

US Patents

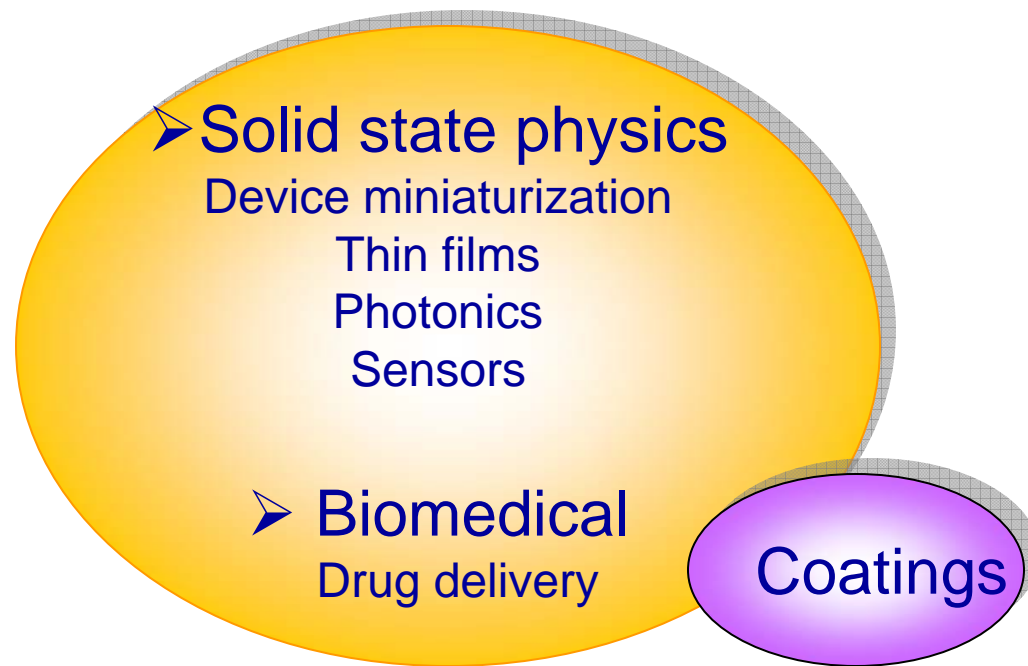
1991-1995 – 4000

2001-2005 – 17,000

...and many more

Nanotechnology

Engineering and Technical contributions to-2008



**Nanotechnology has fueled vigorous research
and development in overlapping areas.**

Why Nano?

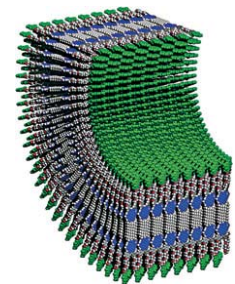
- ◆ Control of coating composition on a molecular level
 - Well-defined composition: “bottom up”

Major areas of Impact

- Barrier
- Corrosion
- Antimicrobial
- Self-cleaning
- Superhydrophobic

↓
-More predictable behavior

↓
-Novel properties



Cross section of
a nanotube
photoconductor

Nanotechnology in Coatings (--to 2008)

Technology	Material	Application	Time-to-market
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Nano-particulate
Coatings

%Effort 95

Polymeric
Nano-materials

%Effort 5

ZnO, Al₂O₃
Ce (III)
Ceramics
Silver, Aluminum
Teflon™
Aniline/Polypyrrole

Supramolecules
Dendrimers &
Hyperbranched
Hybrid

Exterior Automotive
Corrosion control
Fuel cells
Glass coating
Self-cleaning
Super barriers
Drug eluting

Topcoat
Corrosion control
Aerospace component

Current-3 yrs
Current-5 yrs
1-5 yrs
Current-3 yrs
Current-3 yrs
Current-3 yrs
1-3 yrs

2-3 yrs
2-5 yrs
2-3 yrs

Nanotechnology-based Coatings

Materials and Applications



Corrosion Control

Strategies for Corrosion Control by Coating

Protect metal from:

- Oxidation and dissolution
- Prevent electrolyte from reaching the metal surface or keep the concentration at a low level
- Limit water and oxygen transport to the metal
- Interfere with the corrosion reaction
- If corrosion does begin, prevent or reduce its spread

Strategies for Corrosion Control by Coating

Successful Impact

- Cost effective
- Safety
- Material compatibility
- Storage stability

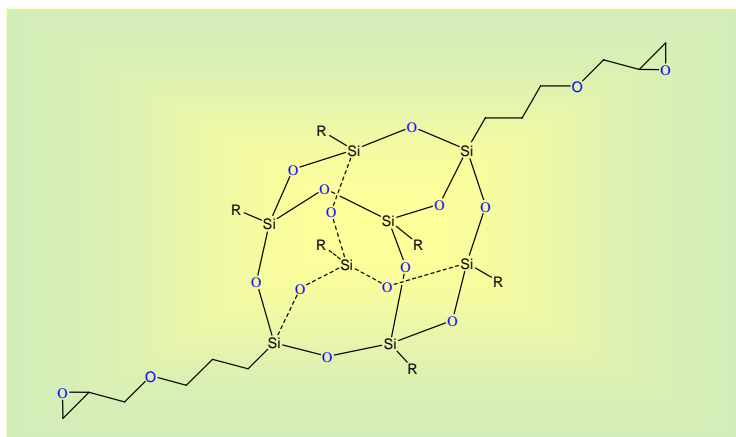
In formulating a coating, one usually, makes certain compromises.

Nanotechnology Approaches

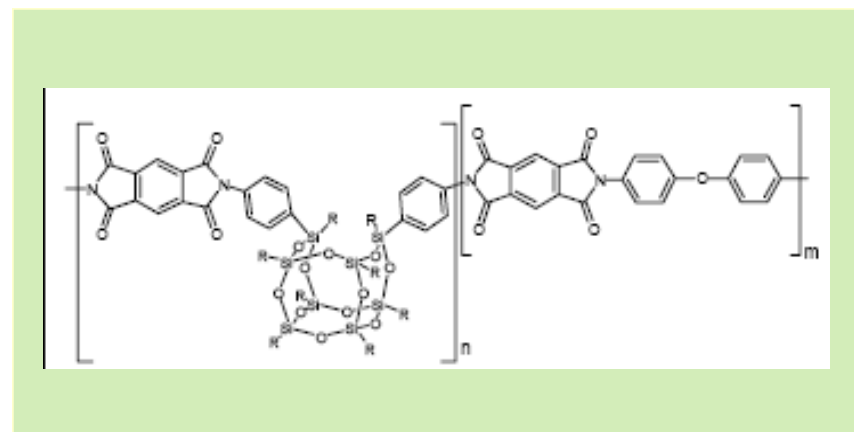
- Conventional Polymers
- Sol-gel Technology
- Inherently Conductive Polymers
- Stimuli responsive/Smart coatings

Nanotechnology-based Corrosion Control Coatings

- Polymer nanocomposite coatings, Al_2O_3 , Fe_3O_4 , $\text{Ce}(\text{NO}_3)_3$, etc.
wang Y., et al, *Wear*, **260**, 976-983, 2006.
- Epoxy systems with dispersed polyaniline nanoparticles
Wessling, B and Posdorfer, J., *Synth. Met.*, **102**, 1400-1401, 1999.
- Fluoro- and silicon/silicone modified polymers
- Organic-inorganic hybrid polymers

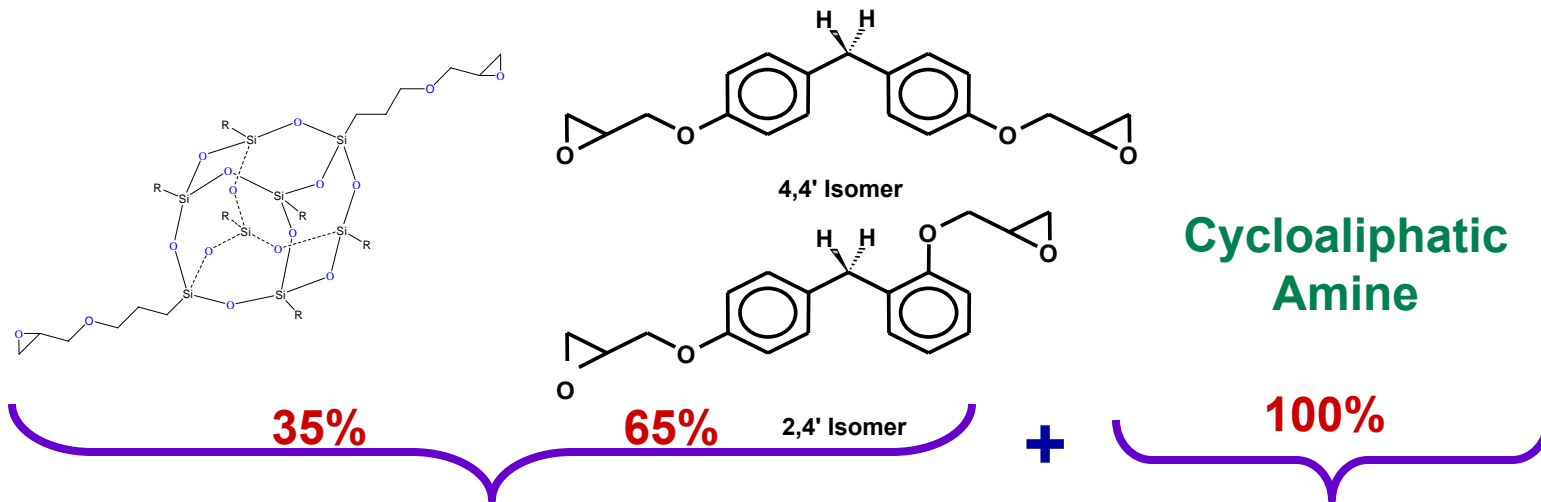


Baghdachi, et al, *Smart Coatings*, 2008

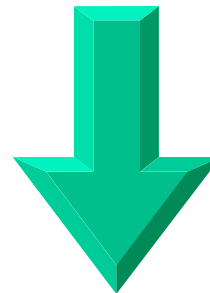


Hopkins, A, The Aerospace Corporation

Nanotechnology-based Corrosion Control Coatings



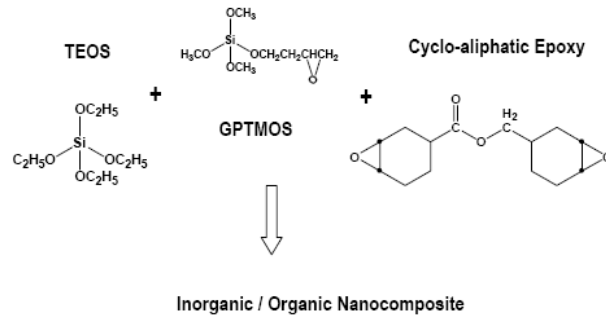
ASTM B117 testing: A, no POSS;
B, with POSS after 680 hrs



Corrosion resistant
epoxy primer base

Sol-gel Technologies

Self-Assembled Nanophase Particle technology
“SNAP” can produce thin diffusion barrier coatings

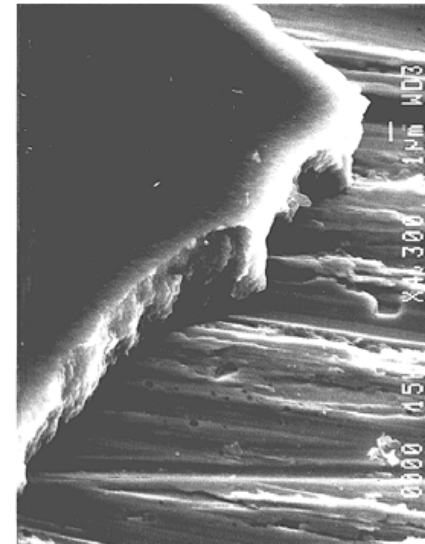


Advantages

- Barrier properties
- RT process

Limitations

- Porosity
- Crack formation
- High bake temp.



Sol-gel silica coating, 3 μm thick on high temperature alloy

<http://www.solgels.com/>

Sol-gel Technologies

Technology Improvements

- Corrosion inhibitor additives

Zheludkenich, M., et al *Surf. Coat Technol.*, **200**, 3084-3094, 2006

Zheludkenich, M., et al *Electrochim Acta*, **51**, 208-217, 2005

Ferreira, M., et al *Electrochim Acta* **49** 2927-2935, 2004

- Barrier property improvement

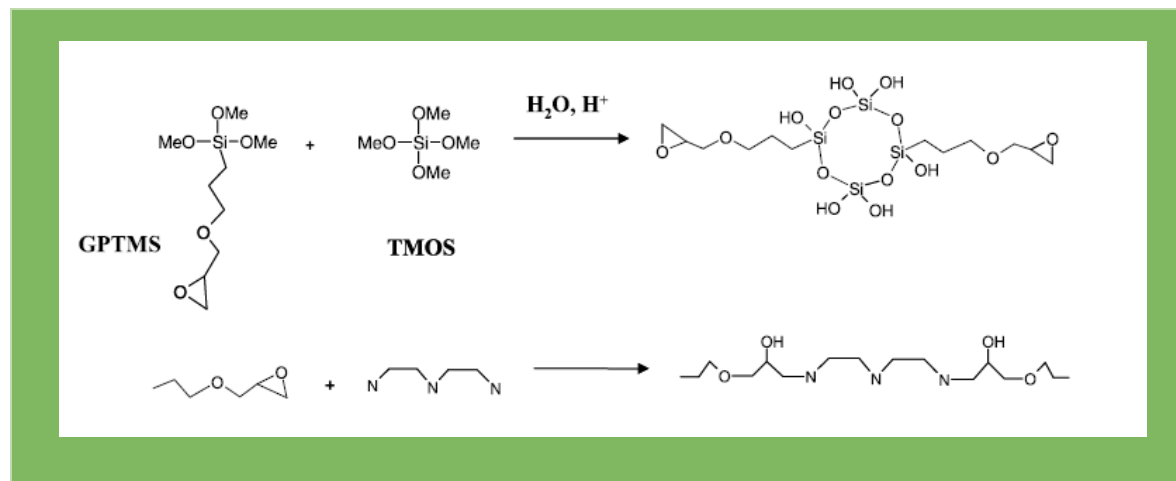
- Khramov, A., et al *Prog. Org. Coat.* **47**, 207-213, 2003

ZrO_2

Ce^{+3}

La^{+3}

Aminosilane
crosslinker



Sol-gel Technologies

Technology Improvements

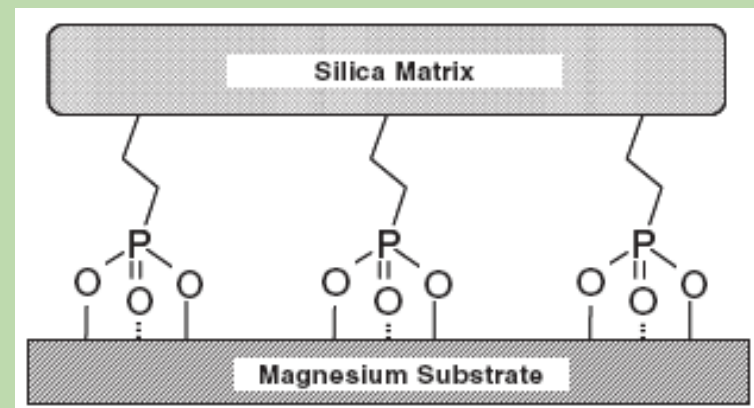
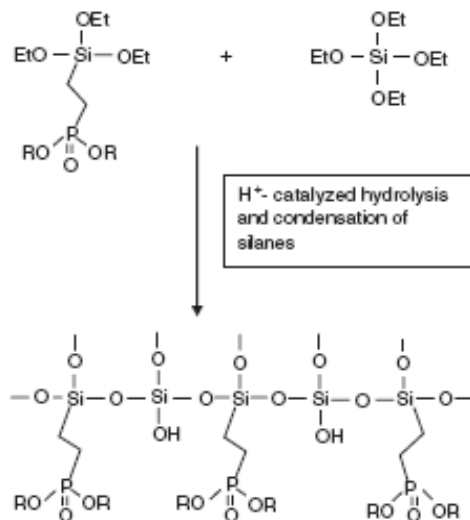
- Self-healing effect

Khramov, A., et al *Thin Solid Films.*, **483**, 191-196, 2005

Aparicio, M., et al *Corros. Sci.*, **50**, 1283-1291, 2008

Kendig, M., *Prog., Org., Coat.*, **47**, 183-189, 2003

- Functionalization



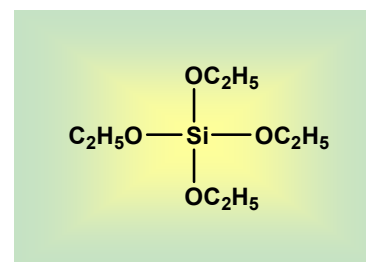
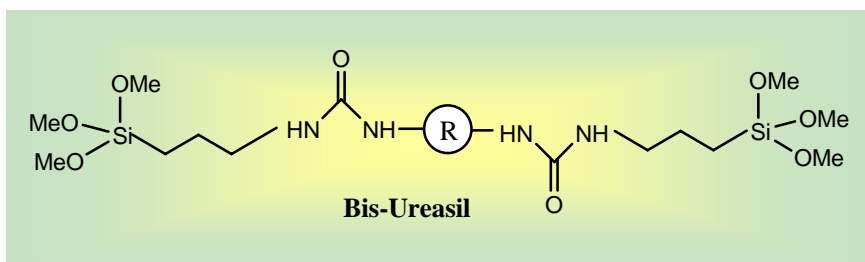
Khramov, A., et al *Thin. Solid. Films.* **514**, 174-181, 2006

Organic
Corrosion
Inhibitors

Ce (III)

Polyaniline

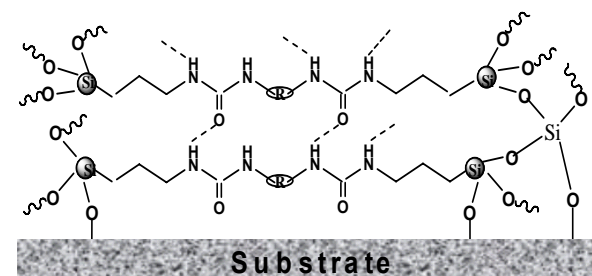
Hybrid Organic-Inorganic Sol-gel Coating



10 P
Alodine™
1200S
1000 hrs

0 P
Control
24 hrs

7 P
CRI
Sol-gel
1000 hrs

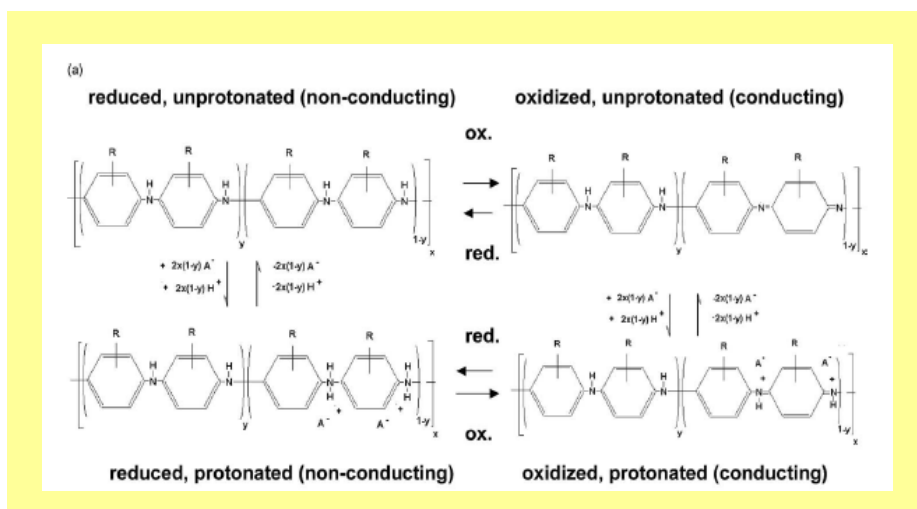


Mannari, V., et al, Eastern Michigan University

Evaluation: as per SSPC – Vis 2 (Pinpoint rusting standard)

Inherently Conductive Polymers

Coatings containing polyaniline in various doped or undoped states increase the corrosion resistance



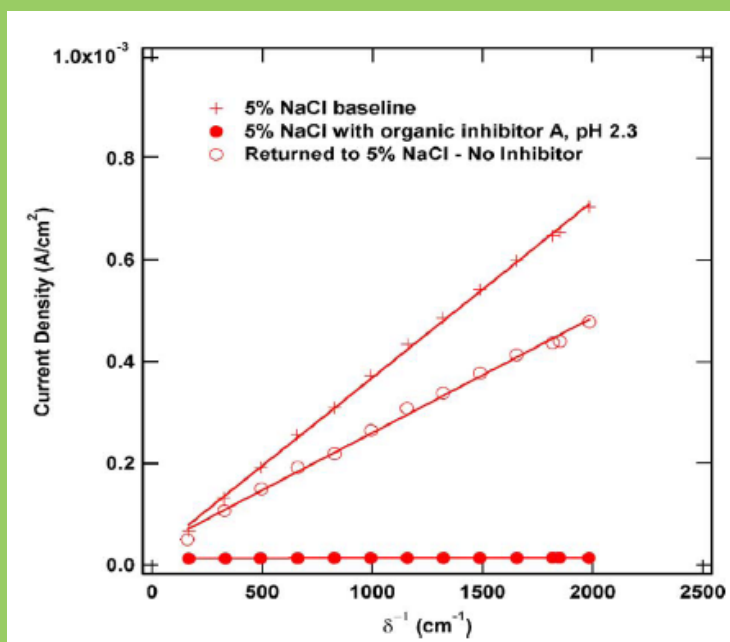
Polyaniline pigmented coatings on steel are highly corrosion resistant in both neutral and acidic media

Talo A. et al, *Synth. Met.* **102** 1394-1395, 1999

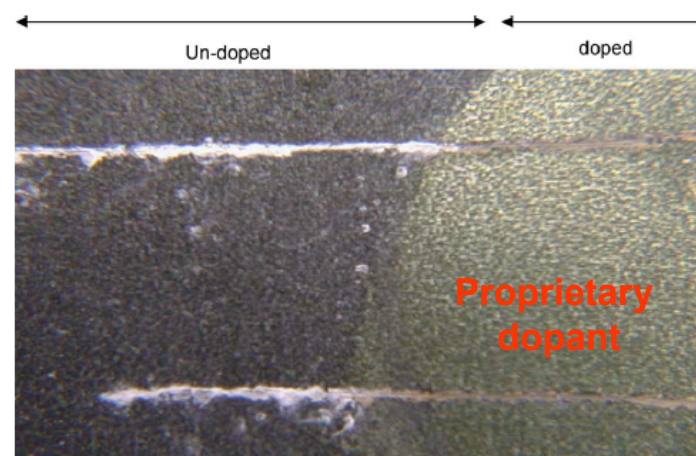
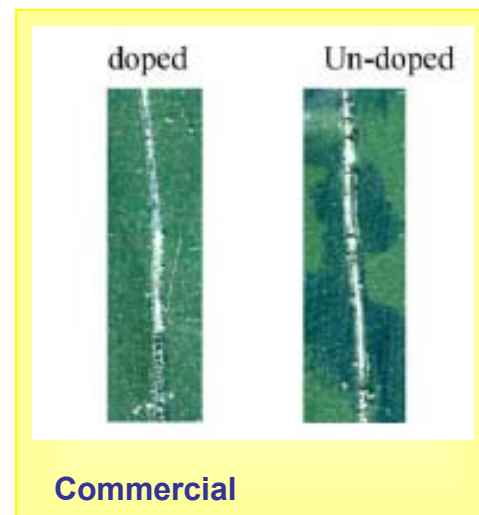
Azim, S., et al. *Prog. Org. Coat.*, **56**, 154-158, 2006

Holness, R. et al. *J.Electrochem. Soc.*, **152** (2)73-81, 2005

Inherently Conductive Polymers “Self-healing”

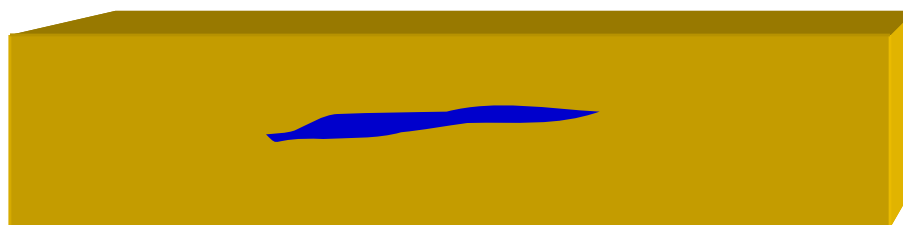


Plot of current density vs. inverse length
for a Cu RDE in aerated 5% NaCl



Self-healing Coatings

- Self-healing in most polymeric systems is achieved by certain morphological tuning or by incorporating stimuli responsive functional materials within the matrix
- Self-healing materials, when damaged, are designed to sense failure, and respond to restore structural integrity



Baghdachi, J., ACS Symposium Series 964, 2008

Self-healing Coatings

➤ Healing Mechanisms

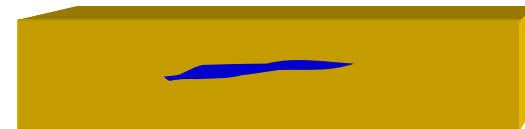
Mechanical forces

S. White, et al. Univ. of Il



Elements of weathering

J. Baghdachi, et al. EMU

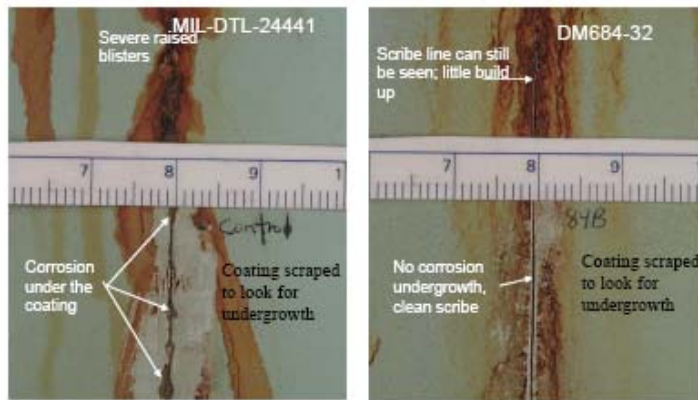
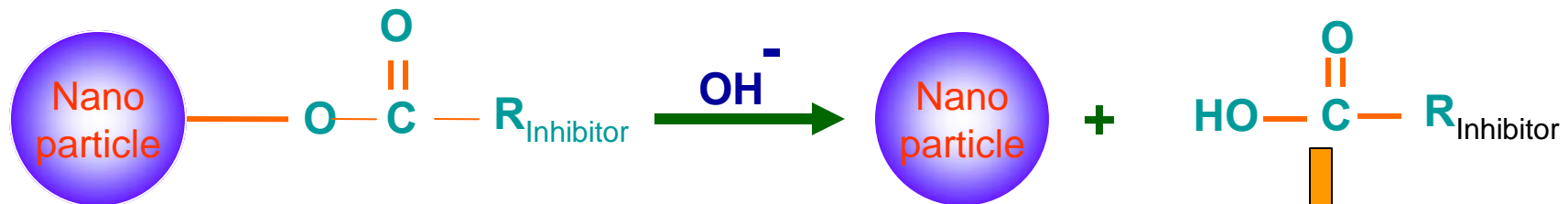
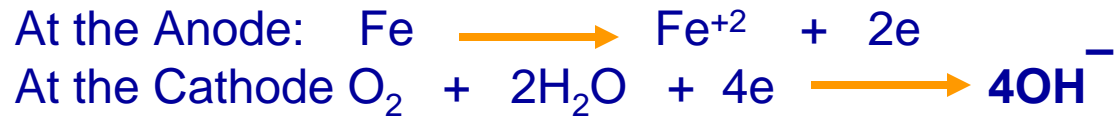


Corrosion by product

L. Calle, et al, NASA



Nanomaterials as Corrosion Inhibitor Components



➤ Corrosion undergrowth in Coating MIL-DTL-24441

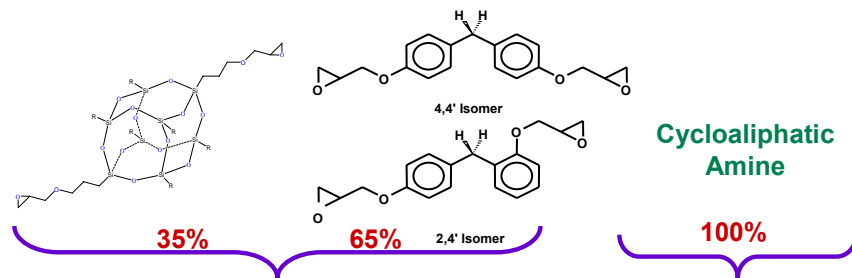
➤ No corrosion under coating with TDA Coating [TDA Research](http://www.tda.com)

Cook, R. TDA Research, www.tda.com

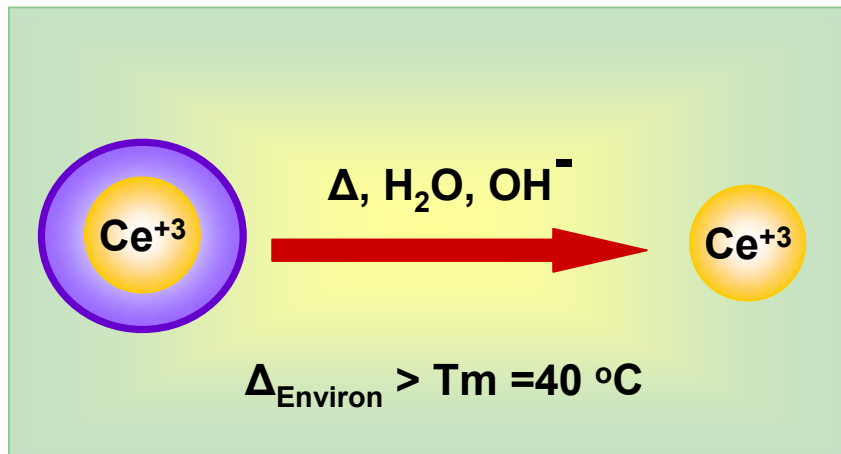
Film former
 Metal

Nanotechnology-based coatings

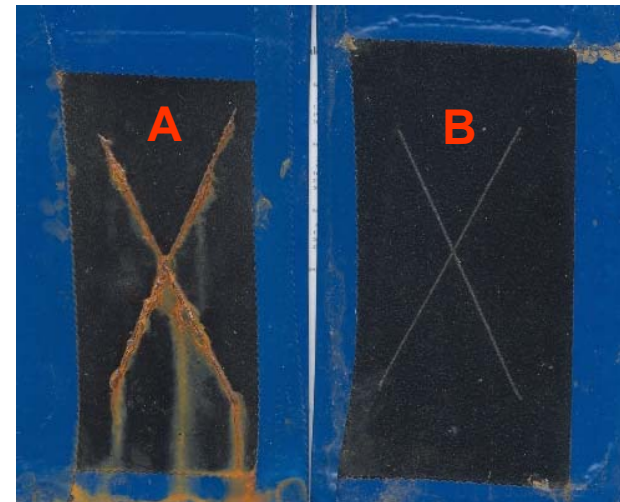
Corrosion resistant Hybrid Organic-Inorganic Coatings



+



Coatings Research Institute



ASTM B117 testing: A, control;
B, with healing agent after 960 hrs

Nanotechnology-based coatings

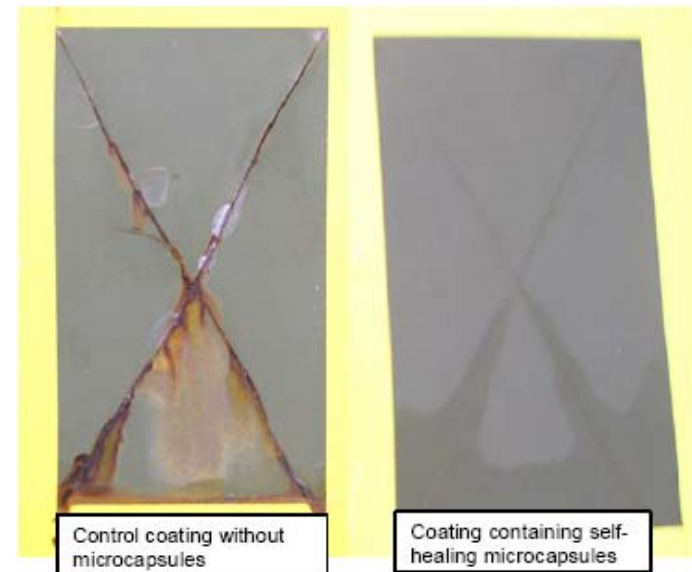
Self-healing Coatings for Corrosion Control

Phenolic varnish plus corrosion inhibitors

Stephenson, L, et al *US 2008/0152815*

Air drying triglyceride plus corrosion inhibitor.

Koene, B., et al, *Proc. Self-healing Conf. 2007*



Luna corporation

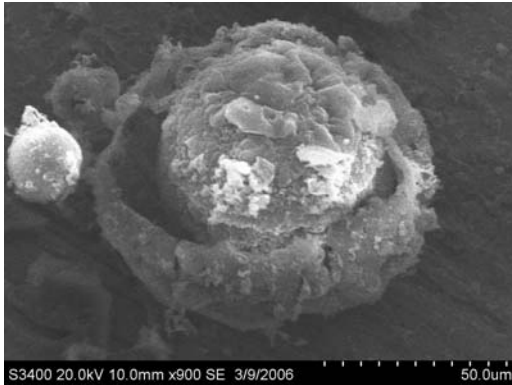
Stimuli Responsive Coatings

Self-healing is triggered by the elements of the weather

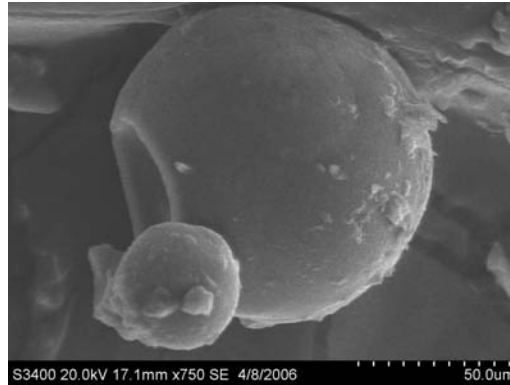


The factors that cause the most damage to the coating also initiate self-healing process.

Approach



Microcapsule with
Bisphenol A epoxy

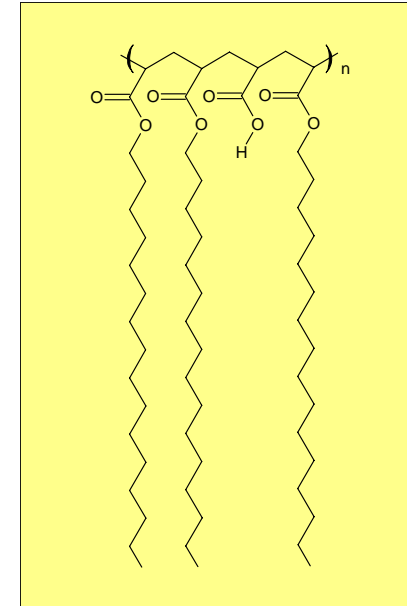


Microcapsule with
Ketimine

Selected SEM images of various microcapsules

Matrix composition:

Bisphenol F resin
Cycloaliphatic amine



Schematic representation of
chemical structure of shell polymer

Self-healing Coatings

Methods and Mechanisms

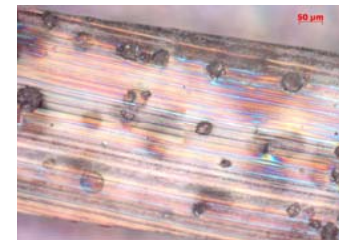
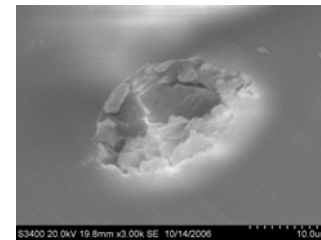
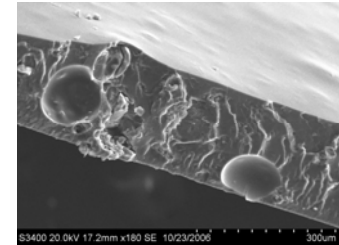
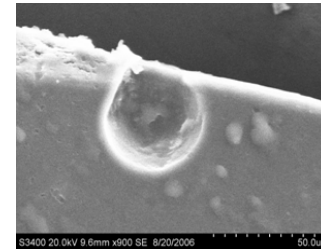
Microcapsule rupture and healing agent release is triggered by:

$T > T_m$

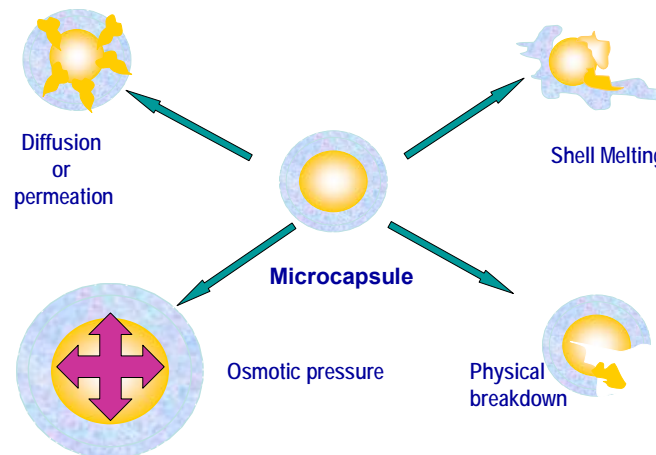
Diffusion through porous shell

Diffusion of water

Osmotic pressure

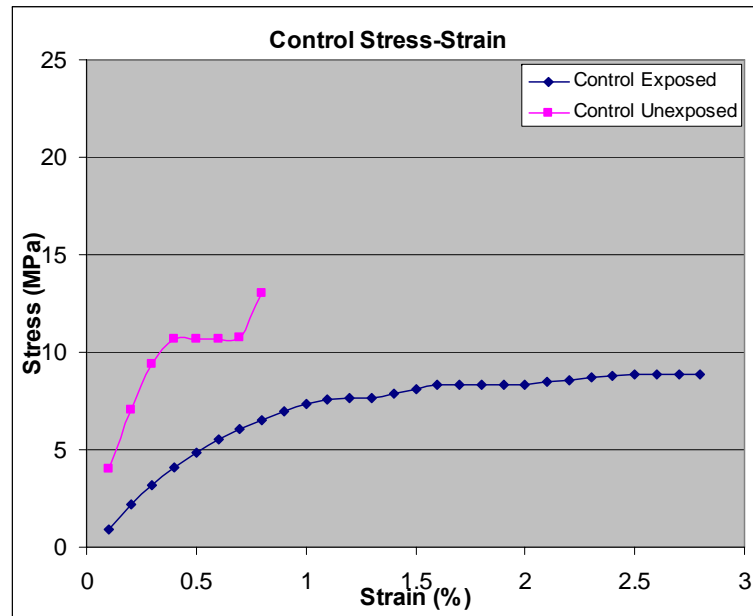


SEM and optical microscopy images of cross-section of self-healing coating

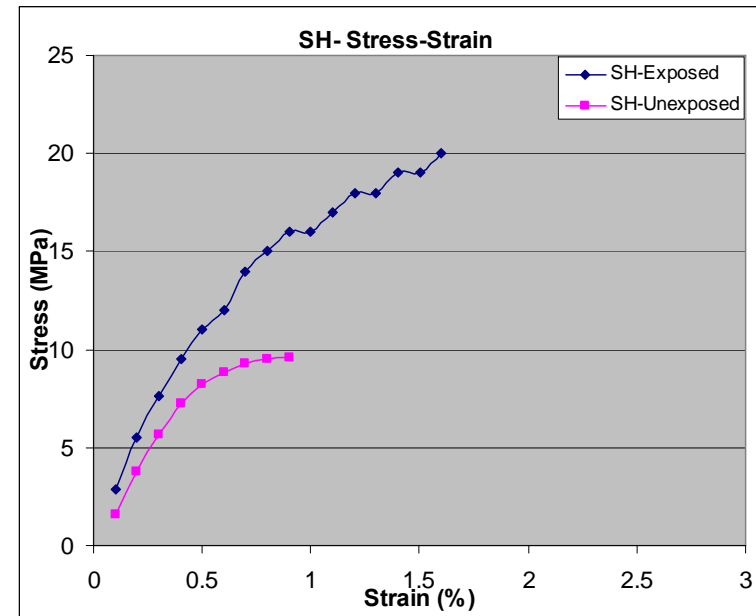


Self-healing Coatings

Dynamic Mechanical Analysis: Stress/Strain



a



b

Control without healing agent (a), -■- Control unexposed, -◆- Control exposed at 65-70% RH, 40-45 °C; sample with healing agent (b), -■- SH-unexposed, -◆- SH-exposed.

Self-healing Coatings

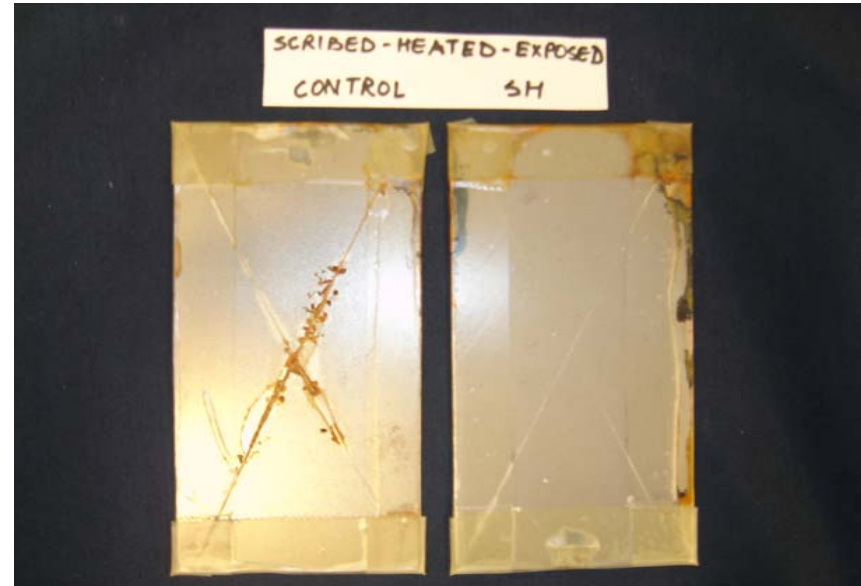
Corrosion Testing, ASTM B117

Objectives

- Confirm self-healing function
- Confirm corrosion resistance improvement
 - Scribed/Exposed, (XE)
 - Scribed/Heated (40 °C/10 min)/Exposed, (XHE)
 - Heated 40 °C/Scribed/Exposed, (HXE)
 - Scribed/E100 hrs/ Heated10 min/Exposed, (XEHE)

Self-healing Coatings

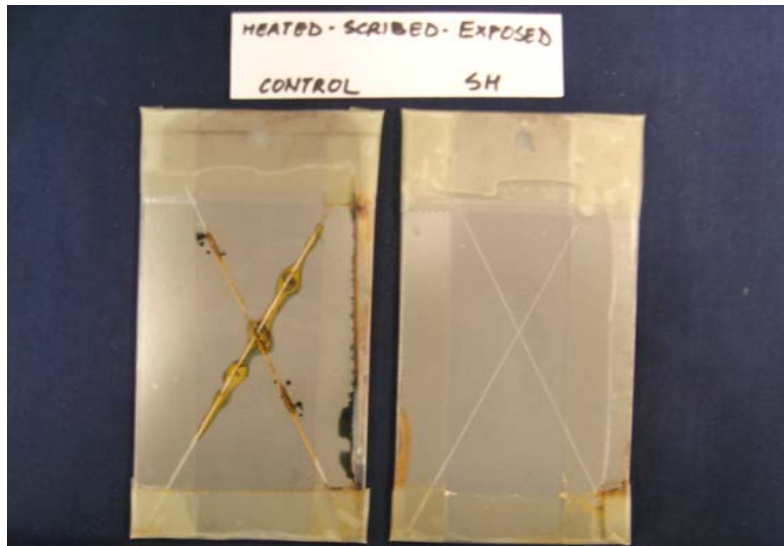
ASTM B117



- Self-healing samples showed no corrosion at 666 hrs
- Control samples corroded after 480 hrs
- Evidence of corrosion at 684 hrs on self-healing samples

Self-healing Coatings

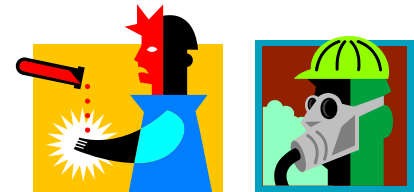
ASTM B117



- Control with or without heat treatment fails corrosion testing
- Heat treatment of damaged coating with healing agents enhances corrosion resistance

Nano Safety

- ◆ **Nanotechnology**, like any new technology, comes with risks
- ◆ Nanomaterials may possess the toxicity of both the bulk forms and the activity and interactions of nano-sized chemicals
- ◆ Increased surface-to-volume ratio of nanoparticles may result in:
 - Ingestion through cell membrane
 - Sensitivity to shape of nanoparticles
 - Adhesion to cell surface



Summary

- ◆ The revolutionary properties of nanomaterials provide evolutionary properties to coatings
- ◆ Nanotechnology approaches have resulted in coatings with improved adhesion and barrier and corrosion resistance
- ◆ Research and development in coatings has been fueled by nanotechnology.
- ◆ Nano-engineered and smart coatings provide the basic function of coatings and achieve results that cannot be attained in any other way.

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